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Answer: $\rho = 737 \text{ kg/m}^3$; $1 \text{ gal} = 3.788 \text{ L} = 3.788 \times 10^{-3} \text{ m}^3$. The mass of a ^{235}U nucleus is $235m_p$, then. (a) for 1 gallon, $m = \rho V = (737 \text{ kg/m}^3)(3.788 \times 10^{-3} \text{ m}^3) = 2.79 \text{ kg} = 2.79 \times 10^3 \text{ g}$. So, $(1.3 \times 10^8 \text{ J}) / (2.79 \times 10^3 \text{ g}) = 4.7 \times 10^4 \text{ J/g}$. (b) 1 g contains $(1.00 \times 10^{-3} \text{ kg}) / (235m_p) = 2.55 \times 10^{21}$ nuclei.

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