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ANSWER: Correct Chapter 9 Reading Question 2 Part A Iron reacts with oxygen in the air to produce rust (Fe 2 O 3) according to the reaction shown below. $4 \text{ Fe} + 3 \text{ O}_2 \rightarrow 2 \text{ Fe}_2 \text{ O}_3$ How many moles of rust can be formed if 1.8 mols of iron are present? Hint 1. Begin with your given number of moles.

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