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Web Activities UNIT 1: ORGANIC CHEMISTRY Chapter 1 – Organic Compounds Section 1.2 questions. p. 23  
Question 5. a. Use the information in Table 3, to plot a graph showing the relationship between the number of carbon atoms and the boiling points of the alkanes. Describe and propose an explanation for the relationship you



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and subject areas that students may need to review. • Should weaknesses or needs be identified, you may want ... s. 12 — — — 3 7) ...

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An organic compound consists of molecules made from carbon, hydrogen

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and oxygen atoms. There can be other atoms – for example, nitrogen, sulfur.

1.19 1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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(a) In CO, there is one carbon atom for every oxygen atom (or the ratio of C to O atoms is 1:1). (b) In CH

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= 12.00 mmol The amount of NaOH(aq) added is also determined:  $n \text{ NaOH(aq)} =$

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$[\text{NaOH}(\text{aq})] \times V \text{ NaOH}(\text{aq}) = (0.300 \text{ mmol/mL})(18.00 \text{ mL})$   
 $n \text{ NaOH}(\text{aq}) = 5.40 \text{ mmol}$   
Unreacted ethanoic acid =  $n \text{ HC}_2\text{H}_3\text{O}_2(\text{aq}) - n \text{ NaOH}(\text{aq}) = 12.00 \text{ mmol} - 5.40 \text{ mmol}$   
Unreacted ethanoic acid =  $6.60 \text{ mmol}$   
Since  $18.00 \text{ mL}$  of  $\text{NaOH}(\text{aq})$  was added to  $20.00 \text{ mL}$  of ethanoic acid solution ...

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## **Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547**

Chemistry 12 - Chapter 1 Quiz.

True/False. ... Identify the letter of the choice that best completes the statement or answers the question. 6.

Which ... 12. The compound whose structure is shown below. a. can be oxidized to an aldehyde. b. can be



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oxidized to a ...

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